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**Product Bulletin** 

Product Name: Laser LL Product Code: 2343021 Revision Date: June 26, 2020

# Laser™ LL

Laser LL is an acidic liquid concentrate designed to remove lead from brass surfaces that will meet drinking water. This will allow the user to pass NSF standards. Laser LL will also remove oxides of copper and brass. In addition, Laser LL can help eliminate the occasional redness that occurs when processing brass. Continuous filtration of Laser LL will allow the user to remove insoluble lead salts and run a steady state operation.

#### **Features & Benefits**

Highly concentrated	Small inventory footprint
Fast acting	Higher productivity
Complete lead removal	Less rejects; higher productivity

## **Operating Conditions**

Laser LL is designed to be used at 10% by volume, but higher or lower concentrations may be used. The unique combination of acids and wetting agents will produce a bright lustrous appearance that is more forgiving than conventional brilliant dips.

The Laser LL should be used at 100°F to110°F, but higher or lower temperatures can be used. Do not exceed 140°F. Lead will be dissolved initially but will become insoluble with time. Simple filtration will remove the lead salts allowing for an extended running of the working bath.

Tanks should be constructed of stainless steel, Polypropylene, Polyethylene, or other acid resistant materials.

Laser LL concentrations can be maintained by simple periodic additions of concentrate, or by a quick titration method which is outlined in the following procedure.

#### **Titration Procedure**

- 1. Pipet 10 mL of cooled bath sample into a 250 mL flask.
- 2. Add about 75 mL water.
- 3. Add 5 drops Phenolphthalein indicator.
- 4. Titrate with 1.0 N Sodium Hydroxide until solution color turns pink.
- 5. Record mL used.

Calculation

Concentration = mL 1.0 N NaOH x 0.943









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Periodic checks of Laser LL should be made to determine and maintain the concentration. If the desire is to run the Laser LL on a continuous basis, the bath may be filtered to remove insoluble lead salts.

### **Waste Disposal**

Cool the working bath to room temperature. With stirring, **slowly** adjust the pH toward neutral with caustic soda, lime, soda ash, etc. Some heat may be generated, so add ingredients slowly and allow to totally dissolve before adding more. When the pH is 4.0-5.0, metal will precipitate. Adjust final pH to 8.0-9.0. Stop mixing and allow solids to settle. The liquid may be decanted to general waste treatment or filter pressed. Make sure all discharge meets permitted discharge requirements of local environmental regulatory authorities.

WARRANTY: THE QUALITY OF THIS PRODUCT IS GUARANTEED ON SHIPMENT FROM OUR PLANT. IF THE USE RECOMMENDATIONS ARE FOLLOWED, DESIRED RESULTS WILL BE OBTAINED. SINCE THE USE OF OUR PRODUCTS IS BEYOND OUR CONTROL, NO GUARANTEE EXPRESSED OR IMPLIED IS MADE AS TO THE EFFECTS OF SUCH USE, OR THE RESULTS TO BE OBTAINED.

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